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Shoji Eguchi,* Yoshio Furukawa, Takanori Suzuki, Kazumoto Kondo, Tadashi Sasaki, Masako Honda, Chuji Katayama, and Jiro Tanaka. Synthesis of Some Novel Functionalized Monoazatricyclic Ring Systems via Intramolecular Cycloaddition of *N*-(Bicycloalkenyl)nitrones.

Page 1899. In supplementary material: the final positional parameters of C(15) in the crystal structure of **14a** (Table III) should read

x/a	y/b	z/c	B_{eq}, Å^2
0.5075 (10)	0.6132 (10)	0.3190 (22)	10.1 (6)

Stanley J. Cristol* and Ellen O. Aeling. Photochemical Transformations. 39. Effects of Ring Substituents and Leaving Groups on Photo-Wagner–Meerwein Rearrangements and Their Ground-State Analogues.

Page 2702. The designations for compounds **16** and **17** are transposed. This occurs in column 1, lines 4 and 5 from bottom, and in column 2, lines 10 and 11 of the text, as well as twice in each of the second and fourth lines of Scheme IV. Further, a CH_3O group is missing from the last structural formula of Scheme IV (same location as shown in **35**).

Ronald A. Forsch and Andre Rosowsky*. A New One-Step Synthesis of Leucovorin from Folic Acid and of 5-Formyl-5,6,7,8-tetrahydrohomofolic Acid from Homofolic Acid Using Dimethylamine–Borane in Formic Acid.

Page 2852, column 2. The statement that the reaction of folic acid with $BH_3{\cdot}HNMe_2$ in formic acid at 0–5 °C affords calcium leucovorin in 64% yield is erroneous. The solid calcium salt isolated in the stated yield was found by C_{18} reversed-phase HPLC (0.1 M NH_4OAc, pH 7.8, with a 0–10% MeCN gradient) to contain ca. 15% leucovorin. The principal component co-eluted with an authentic specimen of 5-methyl-5,6,7,8-tetrahydrofolic acid, whose ultraviolet absorption in H_2O (λ_{max} 285 nm) coincides with that of leucovorin. The agreement of our microanalytical data with an empirical formula consistent with calcium leucovorin was apparently fortuitous. HPLC analysis of the solid isolated from the reaction of homofolic acid confirmed that 5-formyl-5,6,7,8-tetrahydrohomofolic acid was the major component (ca. 50%), but two lesser products and some minor impurities were also present. We are investigating alternative methods of preparation of the desired 5-formyl derivatives via one-pot synthesis.

B. C. Gilbert, D. Griller,* and A. S. Nazran. Structures of Diarylcarbenes and Their Effect on the Energy Separation between Singlet and Triplet States.

Page 4740, Table II. The E value for carbene IV should read 240 MHz and not 440 MHz.

Bong Rae Cho,* Jong Chan Yoon, and Richard A. Bartsch. Reactions of *N*-Halobenzylalkylamines with Sodium Methoxide in Methanol.

Page 4944. Line 14 in Discussion section, "entropies" should read "enthalpies".

Page 4944. Line 17 in Discussion section, "enthalpic" should read "entropic".

Page 4945, Table V. The $\Delta H^\ddagger$ and $\Delta S^\ddagger$ values are miscalculated. The correct values are $\Delta H^\ddagger = 14.2$ kcal/mol, $\Delta S^\ddagger = -19.9$ eu for $ArCH_2N(Cl)CH_3$, and $\Delta H^\ddagger = 14.4$ kcal/mol, $\Delta S^\ddagger = -13.7$ eu for $ArCH_2N(Br)CH_3$, respectively.

Girma Biresaw, Clifford A. Bunton,* and Gianfranco Savelli. Indicator Deprotonation in Micelles of a Hydroxyethyl Surfactant.

Page 5375. The nitro group of 5-nitroindole was misplaced in structures BH and B$^-$.

Jaime J. Cornejo, Kent R. Larson, and G. David Mendenhall*. Applications of Di-*tert*-butyliminoxyl Radical to Organic Synthesis. Oxidation of Amines to Imines.

Page 5382. The name of the second author is Kent R. Larson, not Kent D. Larson.

Tadashi Kometani,* David S. Watt,* Tae Ji, and Tony Fitz. An Improved Procedure for the Iodination of Phenols Using Sodium Iodide and *tert*-Butyl Hypochlorite.

Page 5386. The procedure for the preparation of 4-azido-2-hydroxybenzoic acid is missing the following statement which appears after the preparation of the diazonium salt: To this solution was added 5.09 g of sodium azide in 35 mL of water dropwise over a 15-min period, and the resulting mixture was stirred for 60 min at 0 °C.

B. A. Hess, Jr., C. S. Ewig, and L. J. Schaad*. An ab Initio Study of Cyclobutadiene Dianion and Dication.

Page 5871, column 1. The penultimate paragraph beginning with "The structure of $C_4H_4^{2+}$..." is nonsense and should be deleted. Kos and Schleyer calculated $C_4H_4Li_2$, a cyclobutadienyl dianion ion triplet, in D_{4h} symmetry. We thank Professor Paul von R. Schleyer, Erlangen, for bringing this error to our attention.

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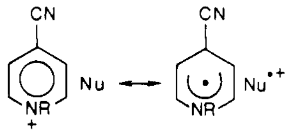
Francesco Minisci,* Elena Vismara, Giampiero Morini, Francesca Fontana, Silvio Levi, Macro Serravalle, and Claudio Giordano. Polar Effects in Free-Radical Reactions. Selectivity and Reversibility in the Homolytic Benzylation of Protonated Heteroaromatic Bases.

Page 479. The nucleophile, Nu, was omitted from eq 12. The correct equation is



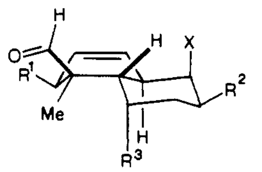
Horst Kessler,* Wolfgang Bermel, Christian Griesinger, Peter Hertl, Edger Streich, and Anton Rieker. Assignments of the Carbon Resonances in *tert*-Butylated 2-Naphthols by the Two-Dimensional C-Relayed H,C-COSY Technique.

Page 600, right column, line 10, should read $^1J_{CC}/P^\pi$ instead of δ/P^π. Line 28 and 29 should read $^1J_{CC}/P^\pi$ instead of π/P^π.

Acknowledgement: Support by the Deutsche Forschungsgemeinschaft and the Fonds der Chemischen Industrie is gratefully acknowledged.

James A. Marshall,* James E. Audia, and Jonathan Grote. Acyclic Stereocontrol in Catalyzed Intramolecular Diels–Alder Cyclizations of 4-Methyl-2,8,10-undecatrienals.

Page 1156. The structure of III in Figure 2 should appear as follows:



III